

**INVESTIGATION OF POLYBROMINATED FLAME RETARDANTS
AND POTENTIALLY TOXIC METALS IN SOIL, SEDIMENT AND WATER
OF THE OBAFEMI AWOLOWO UNIVERSITY DUMPSITE
AND ITS RECEIVING STREAM.**

BY

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CERTIFICATION

This research work by Godwin Oladele Olutona was carried out under our supervision in partial fulfillment of the requirements for the award of Ph.D. Degree in Chemistry of the Obafemi Awolowo University, Ile-Ife, Nigeria.

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DEDICATION

This research work is dedicated to my biological father, **Mr. Stephen Omotayo Oguntona**, who has always encouraged me to pursue education to the peak. To **MANY** that is **UNKNOWN** in the academic profession but are **PRECIOUS GEMS!** To those who have **STRUGGLED** and **SUFFERED** in this noble **PROFESSION**. I salute your courage with deep dedication.

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ABBREVIATIONS AND ACRONYMS

| | |
|-----------------|---|
| ACC | American Chemistry Council |
| ANOVA | Analysis of Variance |
| ATSDR | Agency for Toxic Substances and Disease Registry |
| BFR | Brominated Flame Retardant |
| BSEF | Bromine Science and Environment Forum |
| CDI | Chronic Daily Intake |
| CF | Contaminant Factor |
| DCM | Dichloromethane |
| EFSA | European Food and Safety Authority |
| GC | Gas Chromatography |
| EI ⁺ | Electron impact ionization |
| IARC | International Agency for Research in Cancer |
| ICP-OES | Inductive Couple Plasma-Optical Emission Spectrometry |
| IPCS | International Programme on Chemical Safety |
| MRL | Minimal Risk Level |
| M/Z | Mass to Charge Ratio |
| OCP | Organochlorine Pesticides Residue |
| PBDEs | Polybrominated Diphenyl Ethers |
| PCA | Principal Component Analysis |
| RAIS | Risk Assessment Information System |
| USEPA | United States Environmental Protection Agency |
| WHO | World Health Organization |

ABSTRACT

ABSTRACT

This study investigated the occurrence and levels of polybrominated flame retardants and potentially toxic metals in soils, sediment and water of the Obafemi Awolowo University Dumpsite and its receiving stream. It also monitored the temporal, spatial and seasonal distribution of the analytes in the matrices and evaluated the health risk of the potentially toxic metals in soils and sediments of the study area. These were done with a view to evaluating the pollution status of the studied area with respect to polybrominated fire retardants and potentially toxic metals.

Sampling was done on seasonal basis comprising of May- August and November-February, respectively, for wet and dry seasons. Soil samples of the dumpsite and land space towards the stream were collected using an auger at 0-15 cm and 15-30 cm depths and 50 m away from each other. Water and sediment samples were collected from the receiving stream at six different points that included the upstream and downstream points. Extraction of polybrominated diphenyl ethers (PBDEs) from water, sediment and soil samples were done using liquid-liquid extraction and soxhlet extraction methods with dichloromethane as the extraction solvent. Clean-up of the extracted samples was done using multi-layer silica gel chromatography. The potentially toxic metals were analysed using Inductive Couple Plasma Optical Emission Spectrometry while Gas Chromatography Mass Spectrometry was employed for the quantification of the PBDEs.

The total mean values of PBDEs in the dumpsite soil ranged from 0.36 ± 0.34 ng/g BDE - 47 to 13.84 ± 28.18 ng/g BDE- 153. Total PBDEs indicated that concentrations in 0-15 cm layer were higher than those in the 15-30 cm layer. The concentration of Σ_6 PBDEs in the stream

water ranged from 0.03 – 0.31 ng/ml while the values in the sediments ranged from 0.83 – 10.43 ng/g with BDE-153 occurring as the dominant congener in all the matrices analyzed. Seasonal variability of PBDEs in the matrices indicated that higher levels were found during the wet season in the dumpsite and the receiving stream. The mean values of potentially toxic metals in the dumpsite soil ranged between 3.12 mg/kg Ni and 15500 mg/kg Al. Geo-accumulation study indicated that the soil of the dumpsite and stream sediment were practically unpolluted with, Fe, Cr, Al and Si; strongly polluted with Zr; while the soil and sediment samples were polluted with respect to Se, As, Th, U and Y. Mean levels of the potentially toxic metals in the dumpsite were higher in wet season than dry season. The mean values (mg/kg) of potentially toxic metals in the sediment of Asunle stream ranged from 0.10 Sr to 7260 Fe. The health risk assessment of most of the potentially toxic metals in the dumpsite soil and sediments of Asunle stream revealed that their chronic daily intake for both carcinogenic and non-carcinogenic effects gave hazard quotient for exposure that exceeded the acceptable USEPA value of 1.0.

This study revealed that the environmental matrices of the area under investigation contained higher levels of PBDEs and potentially toxic metals than recommended by WHO and USEPA.

CHAPTER ONE

INTRODUCTION

1.1 Background to the Study

Advances in polymer science over the past 50 years have led to the introduction of a large number of polymers with different properties and applications. As a result, we are surrounded by a wide variety of polymers that have found applications in clothing, furniture, electronics, vehicles and computers. In fact modern cars contain in excess of 100 kg of various polymers. Most of these polymers are petroleum-based, and hence, are flammable.

Fire is a major cause of property damage, loss and even death throughout the history of mankind (Daso *et al.*, 2013a). To circumvent these phenomenal losses as a result of fire accidents, a group of chemicals known as “flame retardants” have been incorporated into various polymer products to reduce the likelihood of ignition and burning in a wide range of textiles, plastics, building materials and electronic equipment used in commerce and residential homes (Alaee and Wenning, 2002). Brominated flame retardants (BFRs) are chemicals that are added to many consumer products including plastics, electronics, textiles, cushioning foams for furniture, automobile interiors and other materials to prevent fire (WHO/IPCS, 1994, 1997).

The idea of flame retardant materials dates back to about 450 BC, when the Egyptians used alum to reduce the flammability of wood. The Romans (about 200 BC) used a mixture of alum and vinegar to reduce the combustibility of wood (Hindersinn, 1990). Certain ammonium salts were also found to be effective for the protection of precious textiles in the early nineteenth century, a practice that continues today (Daso *et al.*, 2013a).

In the development of traditional building materials like wood and metal with plastics, the previously used inorganic salts could not be applied because they considerably reduce

thermal stability (Vonderheide *et al.*, 2008). The development of halogen-based organic flame retardants was a major advancement as they could be incorporated into plastic substance (Daso *et al.*, 2011a). With the increasing usage of polymeric materials in construction, electronic and computer equipment, global market demand for the use BFRs continues to grow substantially. This scientific advancement has resulted in the production and use of several flame retardants broadly classified into three major groups, namely: brominated flame retardants (BFRs), phosphorous-based flame retardants and inorganic flame retardants (usually magnesium and aluminium hydroxides) (Minnesota Pollution Control Agency, 2008). Flame retardants could further be sub-divided into five major classes: brominated bisphenols, diphenyl ethers, cyclododecanes, phenols and phthalic acid derivatives (European Food Safety Authority, (EFSA, 2011). Among these three groups of chemicals, the halogenated organics and organophosphorous flame retardants are of great concern to the public due to their environmental persistence and toxicity (Daso *et al.*, 2011a).

The halogenated organic flame retardants are generally classified as either chlorinated or brominated flame retardants. The brominated flame retardants (Fig 1.1) are extensively used due to their thermal stability, high bromine content and relatively low cost (Darnerud *et al.*, 2001). The most used brominated flame retardants are polybrominated diphenyl ethers (PBDEs), hexabromocyclododecane (HBCD), tetrabromobisphenol-A (TBBPA) and polybrominated biphenyls (PBBs) (Odusanya *et al.*, 2009). Depending on their mode of incorporation into the polymeric materials, two groups of BFRs exist, namely: reactive and additive BFRs. Reactive BFRs are those that are covalently bonded to the polymers, e.g, tetrabromobisphenol A and its derivatives. Additive BFRs, which include polybrominated biphenyls (PBBs), poly brominated diphenyl ethers (PBDEs) and hexabromocyclododecane (HBCD), are not covalently bonded to

the polymeric materials. However, they are more likely to diffuse out of the treated polymer during their lifetime (de Wit, 2002).

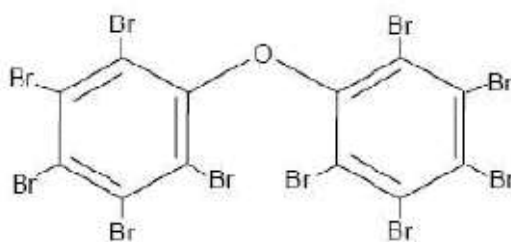


Fig. 1.1: Chemical Structure of Brominated Diphenyl Ethers used in Flame Retardants